PCT







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IMPURITY OF 0.5 %W/W

(57) Abstract

A process for purifying GMP from milk derived starting materials on a production level scale is disclosed wherein the purified GMP has an amino acid composition containing not greater than about 0.5 %(w/w) phenylalanine (PHE). Such a purified GMP product may be used as a dietary supplement for PKU patients. The process comprises contacting a GMP containing feedstock with an anion exchanger under conditions which adsorb the GMP, eluting the GMP, removing the impurities therefrom in a number of alternative ways, and recovering the purified GMP.

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PROCESS FOR ISOLATING GLYCOMACROPEPTIDE FROM DAIRY PRODUCTS WITH A PHENYLALANINE IMPURITY OF 0.5 %W/W

TECHNICAL FIELD

This invention relates to a process for isolating glycomacropeptide (GMP) from a feedstock. More particularly, it relates to the isolation of GMP to a purity under which it has an amino acid composition containing less than 0.5% w/w of phenylalanine (PHE).

GMP is one of a number of names for the peptide split off from κ -casein by the enzymes chymosin and/or pepsin. The peptide is also known as casein macropeptide (CM) or casein derived peptide (CDP).

GMP is found in sweet wheys. GMP carries all of the carbohydrate groups of the parent κ -casein protein. κ -casein is the only casein protein which is glycosylated. Another feature is that when the pH of a solution of GMP is less than 4 the molecular weight is 9000 Da. When the pH is greater than 4, the GMP apparent molecular weight increases to 45000 Da. When concentrating solutions of GMP by ultrafiltration it is preferable for the solution to have a pH >4 unless the membrane cut-off is \leq 10,000 Da. A further feature of GMP is that it does not contain any aromatic amino acids including PHE in its structure.

GMP has a number of potential therapeutic uses as well as having functional properties which make it very useful as an ingredient in food compositions. One important utility is as a nutritional component for use in the diets of persons suffering from phenylketonuria (Marshall S, (1991), Food Res Quarterly, 51, 86-91). Phenylketonurics lack PHE hydroxylase in their metabolic system. Therefore, they are unable to utilise PHE present in foods. This can result in a sufficient accumulation of PHE to cause irreversible mental retardation. In order for GMP to be safe for use in feeding to pheynlketonuries the PHE level should be as low as possible. A representative product specification would require that the PHE level be 0.5% w/w or less and it would be desirable to have a method of isolating GMP to such a low level of PHE impurity on a production level scale.

BACKGROUND ART

Although a number of processes for isolating GMP are known, none of those processes have been shown to be capable of producing GMP having 0.5% wiw or less PHE impurity on a production level scale. One reason for this is that a GMP product of sufficient purity

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has not yet been manufactured. For example, assuming that the contaminant proteins contain an average 4% PHE, then a purity of greater than 88% GMP is required to give less than 0.5% PHE.

For example, EP A291264 discloses an industrial scale process for the purification of GMP from a mixture of α, β and κ caseins whereby the caseins are subjected to enzymatic milk-coagulating treatment to obtain non-coagulate components as an effluent. The effluent is then kept or rendered acidic to form a precipitate and the remaining effluent subjected to a desalting treatment to produce a GMP with 82% purity. No amino acid analysis data of the purified GMP is disclosed. This process is further described in Tanimoto et al, Biosci. Biotech. Biochem., 56(1), 140-141, 1992, which discloses a large-scale preparation of GMP from rennet casein whey. The whey was filtered and the filtrate desalinated and freeze dried to produce GMP powder having a PHE content of 2.4% w/w. The GMP powder was further purified by Q-Sepharose ion-exchange chromatography to give a purified GMP with 0.9% w/w PHE.

GB 2188526 discloses a process for producing a proteinaceous material from milk or casein-containing milk products at pH 4-6 using anion exchange chromatography. No amino acid analysis data of the purified proteinaceous material is disclosed.

GB 2251858 discloses a process for producing GMP from milk raw materials by adjusting the pH of the material to £4, contacting the solution with an anion exchanger, concentrating and desalinating the eluate to give a GMP a 51% purity. This product may be further purified by ultrafiltration according to US 5075424 (below) to produce GMP of 87% purity. No amino acid analysis data of the purified GMP is disclosed.

Outinen et al, Milchwissenschaft 50(10), 570-574, 1995, discloses a process for isolating GMP from cheese whey using an inexpensive polystyrenic strong basic anion exchange resin at pH 5. The PHE content of the purified GMP was 0.9% w/w. Further purification of this product by TCA precipitation gave a PHE content of 0.63-0.79% w/w.

US 5075424 discloses a process for producing GMP from milk starting materials at pH \leq 4 by ultrafiltration. This process relies on the discovery that GMP has a smaller apparent molecular weight at pH \leq 4 than at pH \geq 4 so that only GMP will pass through a 10,000-50,000 molecular weight cut off membrane at pH \leq 4. After ultrafiltration, the pH of the permeate is adjusted to pH \geq 4 and subjected to a second ultrafiltration using a

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membrane with a molecular weight cut-off of ≤50,000 Da to give a GMP retentate which is 82% pure. No amino acid analysis data of the purified GMP is disclosed.

Kawasaki et al, Milchwissenschaft 48(4), 191-195, 1993 discloses a purified GMP product using the method of US 5075424 having a PHE content of 0.6% w/w and a purity of 81%.

Kawasaki et al, Milchwissenschaft 47(11), 688-693, 1992 discloses the same GMP product as above which is fractionated further by analytical chromatography systems (size exclusion chromatology and anion exchange chromatography). The purified GMP products had PHE content of 0.1-0.3%. However, such analytical systems are not suitable for mass production.

US 4042575 and US 4042576 each disclose a process of purifying glycoproteins (including GMP) by double ultrafiltration or flocculation of whey proteins followed by ultrafiltration. No amino acid analysis data for the purified glycoproteins is disclosed.

JP 04243898 discloses a process for producing GMP from cheese whey etc at pH 3-6 by heating, adding ethanol, centrifuging and loading the supernatant onto an anion exchange column and eluting GMP with 0.3M ammonium bicarbonate. This process is also reported in J. Dairy Sci 74, 2831-2837, 1991, where amino acid analysis data is given for the isolated GMP. However, PHE is not included in their analysis.

AU 74081/91 discloses a process for producing GMP from a whey protein concentrate in which the proteins are flocculated, the supernatant concentrated by ultrafiltration and the retentate treated with ethanol to produce a precipitate and a second supernatant. The second supernatant is collected and dried to give GMP powder of 84% purity. No amino acid data is disclosed.

JP 3-294299 discloses a process for the manufacture of GMP from whey by heating a 5-50 wt% solution of whey proteins followed by freezing and thawing. The supernatant is then separated, desalted and concentrated by ultrafiltration. No purity or amino acid analysis data of the recovered GMP is given.

WO 94/159252 discloses a method of producing GMP from whey using ultrafiltration, heat treatment of the retentate at 95°C for 15 minutes, adjustment of the pH to 4-5 filtering

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and collecting GMP from the filtrate. The GMP is 70% pure and the PHE content is 1/3 that of the raw material. The exact concentration of PHE is not disclosed.

EP 0488589 discloses a process for producing GMP by contracting milk raw materials with an ion exchanger, collecting the protein which does not adsorb on the ion exchanger, concentrating and desalting to obtain GMP of 55%-88% purity. No amino acid analysis data is provided

All of the aforementioned processes are aimed at producing enriched GMP fractions and are not necessarily concerned with reducing PHE content of the GMP as they are not concerned with producing diets for phenylketonuries.

According to WO 93/17587, PHE may be removed from proteinaceous material such as whey protein by enzyme hydrolysis and ultrafiltration to remove unhydrolysed protein. The permeate containing mainly amino acids and small peptides is passed through a column of adsorption resin to remove PHE at pH 6-7 (to give a product with 0.3% PHE) or pH 3-5 (to give a product with \(\leq 0.1\% \) PHE). Methods involving enzymatic hydrolysis of proteins such as this one cannot be used to purify GMP since they would degrade it to low molecular weight fractions. They are thus not suitable for producing purified GMP.

Smithers W et al. Food Australia 43(6), June 1991, 252-254 outlines a method of isolating GMP (CDP) from cheese or rennet casein whey by selective adsorption at pH 5 using an anion exchanger. Non-bound proteins are then washed out of the exchanger using water and the GMP eluted with mild acid. The eluate is microfiltered to remove a contaminant and subsequently concentrated by ultrafiltration and dried to give a GMP powder stated to have >90% purity but no PHE analysis data is given.

The present invention provides alternative methods of producing purified GMP having 0.5% w/w or less PHE impurity on a production level scale.

SUMMARY OF THE INVENTION

It is therefore an object of this invention to go some way towards achieving this desideratum or at least to offer the public a useful choice.

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Accordingly, the invention may be said broadly to consist in a process for the purification of GMP characterised in that the purified GMP has a PHE content of not greater than about 0.5%(w/w) which comprises:

5 contacting a GMP containing feedstock with a first anion exchanger under conditions to adsorb said GMP,

eluting said adsorbed GMP from said anion exchanger and removing impurities from said GMP containing eluate by either:

- i) contacting said GMP containing eluate with a cation exchanger in conditions under which said impurities in said eluate are adsorbed onto said cation exchanger, or
- 15 ii) precipitating said impurities in said GMP containing eluate using conditions by means of which said GMP remains in solution, or
 - iii) contacting said GMP containing eluate with a second anion exchanger in conditions under which said impurities in said eluate are adsorbed onto said anion exchanger and,

recovering said GMP from whichever one or more of said steps (i), (ii) or (iii) was used.

In one alternative said conditions in said first anion exchanger are such that said GMP is adsorbed selectively.

In another alternative said conditions in said first anion exchanger are such that both said GMP and other whey proteins in said feedstock are adsorbed and then said first anion exchanger is eluted under conditions which selectively elute said whey proteins and then elute GMP.

Preferably said condition for eluting said whey proteins or said impurities is obtained by adjusting said first anion exchanger to a pH of 4-5 if necessary and then using an appropriate eluent such as 20-60mM sodium chloride.

The whey protein in the cluate may optionally be recovered as an additional product.

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Preferably the feedstock for said first anionic exchanger is a cheese or rennet whey, a UF retentate or a whey protein concentrate produced from a cheese or rennet whey or a whey protein isolate produced from one of these by an anion exchanger.

Alternatively said feedstock has been pretreated to remove substantially all the whey proteins contained therein except GMP.

Preferably said whey proteins have been removed by either heat treating said feedstock or contacting it with a cation exchanger.

In a further alternative said feedstock is derived from a casein or caseinate product prepared by acid precipitation and has been treated with a suitable enzyme to release GMP in solution and precipitate casein.

In one embodiment said GMP containing eluate from said first anionic exchanger is treated by said alternative (i)

In a second embodiment said GMP containing eluate from said first anionic exchanger is treated with said alternative (ii).

In a third embodiment said GMP containing eluate from said first anionic exchanger is treated with said alternative (iii).

In another alternative said GMP containing eluate from said first anionic exchanger is treated with said alternative (i) followed by treatment with said alternative (ii)

In another alternative said GMP containing eluate from said first anionic exchanger is treated with said alternative (ii) followed by treatment with said alternative (i)

In another alternative said GMP containing eluate from said first anionic exchanger is treated with said alternative (ii) followed by treatment with said alternative (iii).

In a still further alternative said GMP containing eluate from said first amonic exchanger is treated with said alternative (iii) followed by treatment with said alternative (i).

Preferably said GMP containing feedstock is contacted with said first amonic exchanger at a pH between 3 and 9.

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More preferably, said pH is between 3.5 and 5.

In one alternative said eluent for desorbing said GMP from said first anionic exchanger contains salt in a sufficient quantity to make the solution approximately 20-200mM and acid in a sufficient quantity to lower the pH to 2-3.

In another alternative said eluent for desorbing said GMP from said first anion exchanger contains salt in a sufficient quantity to make the solution about 175mM and base in sufficient quantity to raise the pH to about 8 to 9.

Preferably, said GMP containing eluate in alternative (i) is contacted with said cation exchanger at a pH between about 1.5 and about 4.5 and at a sait concentration of up to 100 mM.

Preferably said salt is sodium chloride.

Preferably said GMP is neutralised, concentrated by ultrafiltration and/or diafiltration prior to drying.

20 Preferably in said alternative (ii) said conditions are pH 3.5-4.5 attained by adding an acid or base to the GMP containing eluate.

Preferably said acid is hydrochloric acid.

25 Preferably said base is sodium hydroxide.

Preferably after said precipitation step the mixture is centrifuged or microfiltered and the supernatant or permeate is either further processed according to alternative i) or is immediately neutralized, concentrated and dried to produce a GMP powder.

Preferably, in said alternative iii) said second anion exchanger is substituted with amino groups bearing a hydrophobic group. Preferably, said hydrophobic group is selected from the group consisting of C_9 - C_{12} hydrocarbon groups including aromatic hydrocarbons. Preferably said hydrophobic group is a C_8 hydrocarbon group.

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Preferably, the GMP loaded first anion exchanger in the initial step is washed with a dilute salt solution, such as 50mM sodium chloride, prior to said GMP elution to remove impurities therefrom.

- Preferably in said alternative iii), said GMP containing eluate is adjusted to a salt concentration of 125-200mM with a physiologically acceptable salt and the pH is at least 7.
- Preferably the non-adsorbed GMP passing through said second anion exchanger is either further processed according to alternative i) or is immediately neutralised, concentrated by ultrafiltration and/or diafiltration and dried.

In one alternative the GMP from the initial stage or alternative i) is concentrated by ultrafiltration before conducting alternative ii) or iii).

Preferably the process is conducted under conditions which promote a high yield.

In another alternative the invention may be said broadly to consist in a process for the purification of GMP to a PHE content of not greater than about 0.5% (w/w) which comprises:

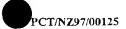
- contacting a GMP feedstock with an anion exchanger under conditions to adsorb selectively said GMP, and
- eluting selectively said adsorbed GMP from said anion exchanger leaving impurities on said anion exchanger, and
 - recovering GMP from the GMP containing eluate.
- In this alternative, these conditions are used to selectively desorb said GMP from the anion exchanger leaving impurities on said anion exchanger in a single step by combining the initial step with alternative (iii).
- Preferably said GMP loaded first anion exchanger in the initial step is washed with a dilute salt solution, such as 50mM sodium chloride, prior to said GMP elution to remove impurities therefrom.

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Preferably said anion exchanger is regenerated by eluting said impurities therefrom after said GMP elution.

Preferably said GMP is selectively eluted with an eluent having a pH of about 8-9 and a salt concentration of about 175mM.

The invention may also be said broadly to consist in a method of purifying GMP from a GMP feedstock substantially as herein described with reference to figure 1 of the accompanying drawings and by reference to any example thereof.

The invention may also be said broadly to consist in GMP purified to contain no more than 0.5% w/w of PHE whenever prepared by a process as defined herein.

More preferably the invention may be said broadly to consist in GMP purified to contain no more than 0.4% PHE whenever prepared by a process as defined herein.

This invention may also be said broadly to consist in the parts, elements and features referred to or indicated in the specification of the application, individually or collectively, and any or all combinations of any two or more of said parts, elements or features, and where specific integers are mentioned herein which have known equivalents in the art to which this invention relates, such known equivalents are deemed to be incorporated herein as if individually set forth.

The invention consists in the foregoing and also envisages constructions of which the following gives examples.

BRIEF DESCRIPTION OF DRAWINGS

One preferred form of the present invention will now be described with reference to the accompanying drawings in which:

Figure 1 is a flow diagram showing the various alternative processes which may be used to purify GMP according to the invention.

Figure 2 is a plot of yield and purity of GMP against pH showing the effect of pH of the GMP adsorption on the yield of GMP and its PHE value.

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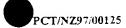


Figure 3 is a plot of time against the absorbance at 214 nm of proteinaceous material eluting off a Mono Q column at pH 8 with an increasing sodium chloride gradient for GMP containing impurities.

Figure 4 is the same plot as figure 3 but for GMP after most of its sialic acid groups have been removed by hydrolysis.

DETAILED DESCRIPTION OF THE INVENTION

Figure 1 shows some of the ways in which alternative methods according to the invention may be combined before producing the final GMP powder as illustrated in the following examples.

In an initial stage of GMP purification a GMP feedstock 16 is contacted with a first anion exchanger 10. Whey proteins (other than GMP) 18 are not adsorbed by the exchanger 10 and are removed after the required contacting time. GMP in the cluate is then removed via line 17 from the anion exchanger 10 which is washed and prepared for another cycle of GMP adsorption.

Eluate in line 17, in one alternative, is fed along feed line 20 into cation exchanger 12. In this case the conditions are adjusted so that impurities are adsorbed and the GMP passes through the cation exchanger 12 along feed line 24 back to line 17 where it is neutralised and passed to an ultrafilter 36 which may be subjected to diafiltration with liquid 37. The retentate is then dried in the drier 38 to produce GMP powder 40.

Impurities are eluted from cation exchanger 12 along with eluate 22. This is referred to as alternative (i).

In another alternative the eluate containing GMP in line 17 is passed to feed line 27. In one embodiment the pH of the solution in feed line 27 is adjusted to ≥4 and subjected to ultrafiltration in the ultrafilter 26 and the permeate 25 is discarded. The retentate is passed through lines 46, 17, 43 and 51 to a precipitation vessel 50 and adjusted to about pH 4 by the addition of acid or base. The mixture is then passed along feed line 53 to centrifuge or microfilter 52. The supernatant or permeate in line 55 is neutralised and returned to line 17 and treated as described above for alternative (i) to produce GMP powder 40. The precipitate or retentate 54 is discarded. In another embodiment the solution from line 17 is fed along lines 43 and 51 directly to the precipitating chamber 50. This is referred to as alternative (ii).



Alternatively, the GMP containing solution from feed line 17 may be fed along lines 43 and 30 into a second anion exchanger 14. The conditions in the second anion exchanger are such that the GMP containing solution passes through and the impurities are adsorbed. In an alternative the GMP containing solution is first diverted from line 17, along line 27, neutralised and passed into an ultrafilter 26. The permeate 25 is discarded. The concentrated retentate from line 46 is passed along lines 17, 43 and 30 into the anion exchanger 14. The GMP containing solution from second anion exchanger 14 is passed along lines 34 and 17 to ultrafilter 36 which may be subjected to diafiltration with the diafiltration liquid 37. The retentate is then dried in dryer 38 to produce a GMP powder 40. The adsorbed impurities in anion exchanger 14 are eluted and discarded in eluate 32. This is identified as alternative iii) of the process.

Although the above descriptions indicate the preferred order in which purification steps may be used, they can be used in any order.

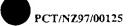
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EXAMPLE 1 - Comparative

A cheese whey WPC powder (sold under the trade mark "ALACEN 392") containing 80% protein was reconstituted at 10% (w/w) solids (GMP concentration 14.2 mg/mL). The pH of this was adjusted sequentially to pH 5.3, 5.1, 4.9, 4.7, 4.2, 4.0 and 3.7 using 5 M hydrochloric acid. At each pH a 50 mL sample was removed and added to 23 g (33 mL) of anion exchanger (QA GiboCelTM, batch no. 1564 (Q2-H), Life Technologies Ltd, New Zealand) stirred for 40 minutes at room temperature while holding the pH constant with 1 M sodium hydroxide. The anion exchanger was then collected on a sintered glass filter, drained and washed with water. The filtrate and washings were combined and made up to 75 mL for analysis of residual GMP by HPLC using a Mono STM (Pharmacia) column. The anion exchanger was transferred to a beaker to which was then added water, 5 M sodium chloride (1 mL) and 5 M hydrochloric acid to give a total volume of 50 mL at pH 2. This was stirred for an hour after which time the exchanger was again collected on the filter, drained and washed with water to give a total volume of 60 mL of eluate (combined filtrate and washings). This was analysed for GMP, α -lactalbumin and β -lactoglobulin by HPLC using ion exchange (Mono STM) and gel filtration (Superdex 75TM, Pharmacia) columns. The solutions of desorbed protein were neutralized, dialysed and freeze-dried. Samples of these were hydrolysed in 6 M hydrochloric acid at 110°C for 23 hours and 110°C analysed for amino acid composition. The % PHE values given herein are as a % of the total amino acids determined in the hydrolysate.

The results are given in Tables 1 and 2 and Figure 2. They show the difficulty of obtaining by ion exchange a GMP product with greater than 88% purity and an amino acid composition with less than 0.5% PHE. This is particularly true using conditions which also favour a high yield of GMP.

TABLE 1

Recovery of	GMP fro	m Chee	se Whe	y WPO			
	pH of protein adsorption						
	3.7	4.0	4.2	4.7	4.9	5.1	5.3
GMP unadsorbed (%)	76	60	49	12	6	2	1
GMP eluted (%)	17	36	49	86	90	91	89
GMP accounted for (%)	93	96	98	98	96	93	90
PHE (% of total amino acids)	-	0.52	0.53	(.69	0.82	1.15	1.93

TABLE 2

		pH of pr	otein ads	orption	
	4.0	4.7	4.9	5.1	5.3
GMP ¹	86.8	82.6	78.8	68.9	46.4
α-lactalbumin ²	1.9	3.7	3.7	4.2	3.8
β-lactoglobulin ²	1.0	1.8	4.4	12.8	27.8
Other ³	10.3	12.0	13.1	14.1	22.0

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- ¹ Concentration in eluate measured using a Mono STM column.
- ² Concentration in eluate measured using a Superdex 75TM column.
- ³ Calculated from the % PHE in the amino acid composition of the final product on the basis of the PHE content of α -lactalbumin and β -lactoglobulin being 4.2 and 3.2% respectively and assuming an average value of 4% PHE for the unidentified impurities.

EXAMPLE 2

A WPC powder (ALACEN 392TM) manufactured from cheese whey was reconstituted at 10% (w/w) solids. The pH of this was adjusted to 4.9 with 5 M hydrochloric acid and 1 litre of it mixed for 40 minutes at room temperature with 460 g (660 mL) of washed and drained anion exchanger (QA GibcoCelTM) while maintaining the pH at 4.9. This mixing was carried out in a vessel fitted with stirrer and a screen across the bottom. At the end of the mixing period the anion exchanger was drained and washed with water.

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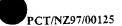
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Water was added to the drained anion exchanger to give a total volume of 1 litre. Sodium chloride (20 mL of 5 M) was added to make the solution approximately 100 mM, and the pH was lowered to 2.0 with 5 M hydrochloric acid. This was mixed for 1 hour at room temperature to desorb GMP and other protein from the exchanger and then drained and the exchanger washed with water to give 1 litre of GMP solution (filtrate and washings) at about pH 2 and 100 mM sodium chloride. The anion exchanger was then ready for use again. A 10 mL sample of the GMP solution was removed, neutralized, dialysed and freeze-dried for PHE analysis. The remainder was passed through a 125 mL column of cation exchanger (SP GibcoCelTM) which had been previously stirred and adjusted to pH 2 prior to settling and packing as a column. The flow rate was 8 mL/min. The last of the protein solution was flushed through the column with 125 mL of water. The column breakthrough and washings were collected and combined to give a total volume of 1.125 L of semi-purified GMP. HPLC analysis using a Mono Q^{TM} column showed that α-lactalbumin, β-lactoglobulin and other material had been removed from the GMP. A 10 mL sample of this GMP solution was removed, neutralized, dialysed and freeze-dried for PHE analysis

The GMP solution was made 200 mM in sodium chloride by the addition of a further 25 mL of 5 M sodium chloride and the pH adjusted to 8.5 with 2 M sodium hydroxide. This was then passed through a 140 mL column of QA GibcoCelTM which had been previously stirred in 200 mM sodium chloride and adjusted to pH 8.5 prior to settling and packing as a column. A flow rate of 10 mL/min was used. The last of the protein solution was flushed through the column with 200 mM sodium chloride. The column breakthrough and washings were collected and combined to give 1.4 litres of purified GMP which was neutralized with 5 M hydrochloric acid. A sample (10 mL) was dialysed and freeze-dried for PHE analysis. The rest was concentrated by ultrafiltration, diafiltration and freeze dried.

Analysis of the crude, semi-purified and purified GMP solutions by UV absorption at 214 nm showed that 83% of the 214 nm absorbing material passed through the SP and QA columns. Much of that lost can be attributed to impurities so that the recovery of GMP through alternatives (i) and (iii) was >83%. The PHE contents of the GMP samples taken at each of the three stages of the recovery and purification process were as follows.

35	Initial Stage	Protein desorbed from QA GibcoCel TM	0.97% PHE
	Alternative i)	Breakthrough from SP GibcoCel ¹³⁴	0.55% PHE
	Alternative iii)	Breakthrough from QA GibcoCel TM	0.08% PHE



EXAMPLE 3

This was the same as Example 2 except that the GMP was desorbed from the first anion exchanger at pH 2 in 75 mM sodium chloride which was then passed through the cation exchange column. This column passed solution (breakthrough solution) of GMP (1.12 litres) was made up to only 175 mM in sodium chloride by the addition of 26 mL of 5 M sodium chloride prior to passage through the second anion exchange QA GibcoCelTM column (140 mL) at pH 8.5.

10 The PHE results were as follows:

Initial Stage	Protein desorbed from QA GibcoCel TM	1.0% PHE
Alternative i)	Breakthrough from SP GibcoCel TM	0.48% PHE
Alternative iii)	Breakthrough from QA GibcoCel TM	0.16% PHE

The overall yield of purified GMP was 70% of the GMP in the WPC powder.

The amino acid compositions of the purified GMP and the theoretical value calculated from the primary structures of the A and B variants (1:1) were as follows:

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TABLE 3

AMINO A	AMINO ACID COMPOSITION OF PURIFIED GMP				
Amino Acid	Theoretical value (%)	Experimental Value (%)			
Asx	7.6	8.28			
Thr	17.5	16 58			
Ser	8.1	7.21			
Glx	18.8	19 66			
Pro	11.8	1141			
Gly	1.0	0.98			
Ala	6.3	531			
Cys	0	0 (0)			
Val	9.0	8 66			
Met	1.9	2.73			
Ile	10.9	10 73			
Leu	1.7	2 50			
Tyr	()	0.08			
Phe	()	0.16			
His	1)	() []			
Lys	5.5	5 32			
Arg	1)	0.29			
Тгр	0	0.0			

EXAMPLE 4

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WPC powder (180 g of ALACEN 392TM) manufactured from cheese whey was reconstituted at 10% solids (w/w) to give 1.8 kg of solution. It was adjusted to pH 4.8 with 2 M sulphuric acid and mixed with 800 g (1.15 L) of QA GibcoCelTM at pH 4.8 for 40 minutes at room temperature. The GMP depleted WPC was then drained from the anion exchanger and the latter washed with water and then resuspended in further water to give a total volume of 1.8 L. This was made approximately 75 mM in sodium chloride by the addition of 27 mL of 5 M sodium chloride, the pH adjusted to 2.0 with 5 M hydrochloric acid and then stirred for 1 hour at room temperature to desorb the GMP. This was drained from the anion exchanger which was washed with water to give 1.8 kg of solution of GMP, pH ≈ 2.0 and 75 mM in sodium chloride.



This GMP solution was passed through a 175 mL column of SP GibcoCelTM at pH 2 and followed by 175 mL of water. The flow rate was 15 mL/min. The first 100 mL of breakthrough from the column was discarded and the remaining 1.875 kg of breakthrough and wash water collected and neutralized to pH 7. Most of this semi-purified GMP solution was heated to 50°C and concentrated 5-6 fold by ultrafiltration using an Amicon spiral wound membrane cartridge (\$1YM10) with a 10,000 molecular weight cut-off Samples of the retentate were further treated in two different ways.

(a) Acidification and clarification

10 The retentate (10 mL) was acidified to pH 4 with 5 M hydrochloric acid and left to stand while some of the precipitated impurities agglomerated. These were removed by centrifugation. The supernatant was neutralized, dialysed and freeze-dried

(b) Anion exchange column

15 The retentate (11 mL) was made up to 170 mM by the addition of 0.22 mL of 5 M sodium chloride and shifted to pH 8.0 with 2 M sodium hydroxide. This was then passed through a 8 mL column of QA GibcoCelTM over 1 hour and flushed through the column with 16 mL of 170 mM sodium chloride. The combined breakthrough and column washings were neutralized, dialysed and freeze-dried.

For comparative purposes a 65 mL sample of the unconcentrated semi-purified GMP was made up to 170 mM sodium chloride, pH 8 adjusted and passed through another 8 mL column of QA GibcoCelTM over 2 hours.

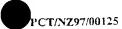
25 The PHE analyses gave the following results:

	Alternative i)	Prior to UF	0.51% PHE
		Post UF	0.55% PHE
	Alternatives ij and ii)	After UF and	
30		precipitation of impurities at pH 4	0.29% PHE
	Alternatives i) and iii)	Final purification by QA GibcoCel column	nn.
		With prior UF	0.30% PHE
		Without prior UF	0.33% PHE

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These results show that precipitation at about pH 4 (alternative ii) may be used to remove impurities as an alternative to passage through an anion exchange column at pH 8-9 (alternative iii).



The amount of PHE containing impurities removed by this method was found to depend on the concentration of the solution and its pH. The impurities are partially soluble at pH 3 - 4.5, and soluble outside this range. Their removal by precipitation is thus aided by concentration by ultrafiltration prior to adjusting to pH 4.

EXAMPLE 5

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A WPC powder (ALACEN 392TM) manufactured from cheese whey was reconstituted at 10% solids. The pH of this was adjusted to 4.75 with 5 M hydrochloric acid and 1 litre of it mixed with 460 g (660 mL) of washed and drained QA GibcoCelTM while maintaining the pH at 4.75 with 2 M sodium hydroxide. After mixing for 40 minutes the GMP depleted WPC was drained through a screen at the bottom of the vessel and the QA GibcoCelTM washed with water and drained.

- The GMP was recovered from the anion exchanger as described in Example 2, except that 75 mM sodium chloride was used (pH 2) and then without any adjustment passed through a 100 mL column of cation exchanger (SP GibcoCel^{1M}) as described in Example 2 to give 1.1 litres of GMP solution (column breakthrough plus wash water of 100 mL)
- The GMP solution was divided into two fractions of 0.55 litres. The sodium chloride concentration in each of these was increased from 68 mM to 175 and 200 mM by the addition of 11.8 and 14.5 mL of 5 M sodium chloride, respectively. The solutions were then adjusted to pH 8.0 with 2 M sodium hydroxide and each passed through a 70 mL column of Q Sepharose Fast FlowTM (Pharmacia) which had been equilibrated to pH 8 in 175 mM and 200 mM sodium chloride respectively prior to packing. The flow rate was 5 mL/min. The last of the GMP solution was washed through the column with 70 mL of 175 mM and 200 mM sodium chloride respectively to give 0.62 litres of column breakthrough and washings. These were neutralized and samples dialysed and freeze dried.



The PHE analysis results from the GMP powders were as follows:

	Initial Stage		0.68% PHE
	Alternative i)		0.41% PHE
5	Alternative i) and iii)	using 200 mM NaCl	0.21% PHE
		using 175 mM NaCl	0.13% PHE

Chromatography of the GMP products on a Mono QTM column showed that only the 0.13% PHE product was free of material eluting from the Mono QTM column after GMP (See Example 11). There were still traces of these impurities in the 0.21% PHE product.

EXAMPLE 6

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This was the same as Example 5 except that alternative i), the intermediate cation exchange column (SP GibcoCel), was omitted. The two final GMP products from the Q Sepharose Fast FlowTM columns gave PHE results of 0.41 and 0.32% respectively. Approximately 0.2% PHE can be attributed to the α-lactalbumin (3.7%) and β-lactoglobulin (1.8%) remaining in this type of product.

20 EXAMPLE 7

This was the same as Example 2 except that the GMP adsorption by the first anion exchanger (QA GibcoCelTM) was carried out at pH 4.2 instead of 4.9. The 1 litre of GMP solution recovered at pH 2 and 100 mM NaCl (mostly glycosylated GMP) was divided into two 500 mL fractions.

(a) Purification through SP and QA Columns

One 500 mL fraction was passed through a 40 mL column of SP GibcoCelTM over 2 hours and then adjusted to pH 8.5 and 175 mM sodium chloride and passed through a 40 mL column of QA GibcoCelTM as previously described.

(b) Purification through QA column only

The other 500 mL fraction was adjusted to pH 8.5 and 175 mM sodium chloride and passed through a 40 mL column of QA GibcoCelTM.

Analyses of samples taken at the various stages gave the following results:

TABLE 4

Purity and Yield of GMP Products					
	PHE (%)	Yield ¹ (%)	Yield² (%)	Sialic Acid (%)	
Initial Stage GMP	0 53	51.0	49.0	•	
Initial Stage GMP after: Alternative i) only	0 38	48.8	47.3	-	
Alternative iii) only	0 22	45.0	-	-	
Alternatives i) and iii)	0 15	44.6	-	12.7	

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EXAMPLE 8

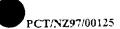
GMP was adsorbed from 1 litre of 10% (w/w) reconstituted WPC (ALACEN 392TM) at pH 4.8 by QA GibcoCelTM using the same techniques as in Example 2. The GMP was then desorbed from the drained and washed anion exchanger by a 2 stage elution as follows:

(a) Elution 1

Water was added to the anion exchanger to give a total volume of 1.1 litres. Aqueous sodium chloride (4 mL of 5 M) was added to make the solution approximately 20 mM, and the pH was lowered to 3.2. After stirring for 1 hour the desorbed GMP solution was drained from the anion exchanger which was washed with 20 mM sodium chloride to give 1 litre of recovered GMP, (mostly aglyco-macropeptide). This was adjusted to pH 3.9 with 2 M sodium hydroxide and clarified by centrifugation. The supernatant (940 mL) was then passed (over 2 hours) through a 120 mL column of SP GibcoCelTM which had been pre-equilibrated at pH 3.9 in 20 mM sodium chloride. It was washed through with water to give 1.12 litres of column passed solution. A sample of this, and other GMP solutions from earlier in the process, were neutralized, dialysed and freeze-dried for PHE analyses. The results are shown in Table 4 below:

¹ Calculated from absorbance readings at 214 nm and using a value of 140 for a 1% solution of pure GMP.

² Analysis of GMP using a Mono STM column. The load solution had a concentration of 14.2 mg GMP/mL.



(b) Elution 2

After the first elution of protein from the QA GibcoCelTM it was again mixed with water to give a total volume of 1.1 litres. The sodium chloride was raised this time to approximately 200 mM by the addition of 20 mL of 5 M sodium chloride and the pH lowered to 3.0 to complete the recovery of GMP (mostly glycosylated) and other protein from the exchanger. After stirring for 1 hour the protein solution was drained from the exchanger and washed out with 200 mM sodium chloride to give 1 litre of recovered GMP. This was adjusted to pH 8.5 with 2 M sodium hydroxide and made up to 250 mM by the addition of a further 10 mL of 5 M sodium chloride. It was then passed through a pre-equilibrated column of QA GibcoCelTM (120 mL) over 2 hours and washed through with 250 mM sodium chloride to give 1.24 litres of solution. Samples were again taken for PHE analysis and the results are shown in Table 4:

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TABLE 5

Purity and Yield of GMP Products					
	PHE	Yield	Sialic Acid		
	(%)	(%)*	(%)		
Eluate 1 Protein desorbed at pH 3.2/20 mM NaC1	1.19	63	2.6		
Supernatant from pH 3.9 centrifugation	0.98	55	-		
Breakthrough from SP GibcoCel™ column	0.31	43	2.6		
Eluate 2 Protein desorbed at pH 3.0/200 mM NaC1	0.51	37	-		
Breakthrough from QA GibcoCel TM column	0.28	33	17.4		

* Yield of 214 nm absorbing material as a % of the total eluted, 10. eluates 1 & 2 = 100%

EXAMPLE 9

(a) A WPC powder (ALACEN 392) manufactured from cheese whey was reconstituted at 10% solids and the pH adjusted to 4.7 with 5 M hydrochloric acid. This solution (1 Kg) was mixed with 460 g (660 mL) of washed and drained QA GibcoCel whilst maintaining the pH at 4.7 with 2 M sodium hydroxide. After 40 minutes the GMP depleted WPC was drained through a screen at the bottom of the vessel and the retained ion exchanger washed with water.

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The settled bed of QA GibcoCel was then washed with 1660 mL of 50 mM sodium chloride solution. Further 50mM sodium chloride was then added to give a total volume in the vessel of 1 litre. This was stirred and the pH adjusted to 8.5-9.0 with 2 M sodium hydroxide. After about 10 minutes 35 mL of 5 M sodium chloride was added to raise the sodium chloride concentration from 50 to 175 mM to selectively elute the GMP. Mixing was continued for a further 30 minutes, whilst maintaining the pH at 8.5. The GMP containing eluate was then drained from the vessel and the bed of QA GibcoCel washed with 2 bed volumes (1320 mL) of 175 mM sodium chloride over 30 minutes to continue the elution of GMP. The total eluate (2 L), containing 71% of the GMP with 0.37% PHE, was concentrated by ultrafiltration, diafiltration and freeze-dried.

The QA GibcoCel was further treated to recover the remaining protein by adding water, adjusting the pH to 2.0 and mixing for a total time of 60 minutes. The eluted protein solution was drained and the ion exchanger washed with water.

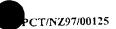
(b) The above process was repeated except that the selective elution was carried out in 150mM sodium chloride instead of 175mM.

The GMP concentrations in the process streams at each stage were determined by analysis on a Mono S column to determine the losses and yield of GMP. These results along with those from amino acid analyses on the final products are shown in Table 6

TABLE 6

Purity and Yield of GMP				
	Yield (%)*	PHE%		
GMP depleted WPC and wash.	13	-		
Salt wash.	4	-		
Selective elusions				
(a) (i) pH 8.5/175 mM NaC1	71	0.37		
(ii) pH 2.0/≈100 mM NaC1	14	1.9		
(b) pH 8.5/150 mM NaC1	58	0.26		

^{*} Based on the GMP present in the reconstituted WPC



EXAMPLE 10

A WPC powder (ALACEN 392) manufactured from cheese whey was reconstituted at 7.5% solids. The pH of this was adjusted to 4.75 with 5 M hydrochloric acid and 1 Kg of it mixed with 350 g (500 mL) of washed and drained QA GibcoCel while maintaining the pH at 4.75 with 2 M sodium hyroxide. After mixing for 40 minutes the GMP depleted WPC was drained through a screen at the bottom of the vessel and the retained ion exchanger washed with water. The bed of QA GibcoCel was then washed with 1250 mL of 50 mM sodium chloride solution to desorb mainly contaminating proteins. This was followed by a water wash of 250 mL which was drained completely from the ion exchange bed.

Water was added to the GMP loaded QA GibcoCel to give a total weight of 750 g. Sodium chloride (9 mL of 5 M) was added to make the solution approximately 60 mM, and the pH was then lowered to 2 0 with 5 M hydrochloric acid. This was mixed for 1 hour at room temperature. The eluted protein was drained from the bed of exchanger which was then washed with water to give 900 mL of GMP solution with a sodium chloride concentration of 50 mM and about pH 2. This was neutralized to pH 9 with 2 M sodium hydroxide and divided in two equal fractions

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- Half the GMP solution (450 mL) was made up to 150 mM sodium chloride by the addition of a further 9 mL of 5 M sodium chloride and passed through a 50 mL column of QA GibcoCel pre-equilibrated to pH 9 in 150 mM sodium chloride prior to packing as a column. A flow rate of 4 mL/min was used. The last of the GMP solution was flushed through the column with 50 mL of 150 mM sodium chloride to give 500mL of purified GMP which was neutralized.
- (b) The other half of the GMP solution (450 mL) was left at 50 mM sodium chloride strength and passed at 4 mL/min through a 50 mL column of DMO cellulose preequilibrated to pH 9 in 50 mM sodium chloride prior to packing as a column. Fifty millilitres of 50 mM sodium chloride was used to flush the last of the GMP solution through the column to give 500 mL of purified GMP solution.

Small samples of GMP containing solutions were kept for GMP analysis by ion exchange on a Mono S column or dialysed, freeze-dried and hydrolysed for phenylalanine analysis by amino acid assays. The results are shown in Table 7.

DMO collulose was a dimethyloetylamino collulose prepared in the laboratory from LIP GincoCalTM using the method of L Matsumoto *et al.*, J Biochem. 37, 535-540 (1980) to prepare an epexy-activated collulose. This was then counted with N₂M-dimethyloetylamine in aqueous ethanol to give DMO collulose with a substitution level of 0.8 metry (55 acq mL).

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TABLE 7

Purity and Yield of GMP				
	Yield (%)*	PHE%		
GMP depleted WPC and wash.	5.0	-		
Salt wash.	5.5	-		
Initial Stage	82	0.76		
Alternative (iii) using				
(a) QA GibcoCel in 150 mM NaC1	71	0.40		
(b) DMO Cellulose in 50 mM NaC1	71	0.28		

* Based on the GMP present in the reconstituted WPC.

EXAMPLE 11

15 Cheese whey WPC powder was reconstituted to give 2 litres of a 10% (w/w) solution. This was heat treated to precipitate the whey proteins except for GMP and centrifuged at 90°C. One litre of the GMP containing supernatant was then adjusted to pH 4.8 and treated with 660 mL of QA GibcoCelTM and the GMP obtained further purified as described in Example 3. Samples (10 mL) were taken at each stage, neutralized, dialysed and freeze-dried. PHE analyses gave the following results:

		PHE %
Supernatant from heat	treated WPC	2.15
Initial Stage	Protein desorbed from QA GibcoCel TM	1.03
Alternative i)	Breakthrough from SP Gibcocel TM	0.34
Alternative i) and iii)	Breakthrough from QA column	0.31

EXAMPLE 12

Most of the GMP samples produced in the proceeding examples were analysed by anion exchange chromatography using a Mono Q PCTM 1.6/5 column coupled to a Pharmacia SMARTTM System. Samples (2 mg/mL) were dissolved in 0.01 Tris buffer at pH 8.0. The Mono QTM column was equilibrated and run under the following conditions using:

Buffer A: 0.01 M Tris, pH 8.0 and Buffer B: 0.01 M Tris, 1 M NaCl, pH 8.0

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TABLE 8

Gradient Table				
Time min	Flow µL/min	%A	%B	
0	80	90	10	
12.5	80	72	23	
18.0	80	50	5:0	
20.0	80	0	1.00	
23.0	80	0	100	
26.0	80	90	10	
31.0	0	90	10	

Figure 3 shows a typical chromatogram obtained for the GMP samples obtained in alternative i) of the process as set out in Examples 2 and 3, i.e. the protein desorbed from the QA GibcoCelTM and passed through the SP GibcoCelTM column at pH 2.

The peaks eluting between 4 and 13 minutes were mainly GMP, i.e. the A and B variants of the aglyco and variously glycosylated forms of the macropeptide obtained from κ -casein. The peaks eluting after 13 minutes were not GMP and these impurities gave rise to significant levels of PHE in the GMP obtained from a single stage extraction of GMP from cheese and rennet casein wheys and retentates by ion exchange. α -lactalbumin, BSA, β -lactoglobulin and these other impurities were removed from the GMP by a multistage process, such that they were no longer detectable by chromatography on a Mono QTM column, then GMP was obtained with < 0.2% PHE in its amino acid composition. Complete removal of the peaks eluting between 13 and 15 minutes was not always achieved resulting in products with 0.2 to 0.4% PHE. Such products have been readily obtained on a 200 litre scale using GibcoCelTM exchangers with retentates and reconstituted WPC powders from rennet casein whey and various kinds of cheese wheys. The distribution and size of the peaks in the 13-15 minute range varied with different whey types but the bulk of them were removed by the processes set out in these examples so that the PHE content of the final product was not affected by these differences.

EXAMPLE 13

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This was the same as Example 3 except that the semi-purified GMP, obtained after passage of the GMP through the SP GibcoCelTM column at pH 2 in 75 mM sodium chloride, was heated for 20 minutes at 80°C to hydrolyse the sialic acids groups off the various glyco forms of the macropeptide.

After cooling, 3 x 360 mL fractions of this hydrolysed solution were made up to 125, 150 and 175 mM in sodium chloride by the addition of 4, 5.8 and 7.6 mL of 5 M sodium chloride respectively. Each of these fractions was then adjusted to pH 8.0 and passed through a 40 mL column of QA GibcoCelTM previously adjusted to pH 8.0 in 125, 150 or 175 mM sodium chloride prior to settling as a column. The flow rate was 3 mL/min. The last of each GMP solution was flushed through the column with the appropriate strength of sodium chloride to give 400 mL of purified GMP solution. Samples of each solution were diluted 1 in 100 with water and the absorbance at 214 nm measured to estimate the GMP concentration using a value of 140 for the absorbance of a 1% solution of pure GMP at 214 nm. The results in Table 3 validate this method of analysis for very pure samples of GMP. Other samples were neutralized, dialysed and freeze-dried for PHE analysis

Figure 4 shows the Mono QTM chromatogram for this hydrolysed GMP and the very large separation now between the two main asialyo GMP peaks and the PHE containing impurities at 13 - 15 minutes. This made it possible to use lower salt strengths for the QA GibcoCelTM column to bind these impurities in alternative iii) without also binding the GMP. The results in Table 9 show that the sodium chloride concentration in the range 125-175 mM is not particularly critical to the performance of the anion exchange column used in this alternative of the process.



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TABLE 9

	РНЕ%	yield%1	sialic acid%2
GMP from initial stage	1.0	-	-
Hydrolysed GMP from alternative i)	0.48	-	-
GMP from 175 mM NaC1 QA column	0.20	74	-
GMP from 150 mM NaC1 QA column	0.14	70	1.2
GMP from 125 mM NaC1 QA column	0.16	66	-

10 Yield of the GMP that was present in the WPC at the start.

² The final GMP obtained in Example 2 (no hydrolysis) had a sialic acid content of 5.4%.

EXAMPLE 14

A WPC powder (ALACEN 392TM) manufactured from cheese whey was reconstituted in demineralised water at 15%(w/v) solids to give 1700 L of reconstituted retentate. The pH of this retentate was adjusted to pH 4.75 with 29 L of 10% HC1 and 160 L of it mixed with 150 L of anion exchanger (QA GibcoCelTM) in a bed of water. The pH was maintained at 4.75 whilst continuing this mixing for 30 minutes at 10° in a vessel fitted with a stirrer and a screen across the bottom. At the end of the mixing period the GMP-depleted retentate was drained from the anion exchanger which was then washed with water.

Demineralised water was added to the washed anion exchanger to give a total volume of 300 L. Cheese salt was added to make the solution approximately 55 mM in sodium chloride and the pH was lowered to 2.0 with 10% HC1. This was mixed for 1 hour at 10°C to desorb the GMP and other protein from the exchanger and then the exchanger was drained and washed to give 250 L of eluate and washings. Nine further cycles were carried out in this manner to give a total of 2500 L of GMP eluate.

The GMP eluate, at pH 2.0 and 55 mM sodium chloride, was passed at 250 L/hr and 10°C through a column containing 80 L of cation exchanger (SP GibcoCelTM) which had been previously stirred and adjusted to pH 2 prior to settling as a column. The eluate was flushed through the column with 80 L of water.

The GMP solution that had passed through the cation exchanger was made 175 mM in sodium chloride by the addition of 11 kg of sodium chloride and then the pH was adjusted to pH 8.5 with 10% sodium hydroxide. This was then passed at 475 L/hr and 10°C through 150 L of QA GibcoCelTM which had been previously stirred in 175 mM sodium chloride and adjusted to pH 8.5 prior to settling and packing as a column. The last of the GMP solution was flushed through the column with 175 mM sodium chloride.

The GMP solution (2800L) that had passed through the anion exchanger was neutralised to pH 6.7 with 1.3L of 10% HC1. It was then ultrafiltered at 10°C in batch mode to VCF 15 using two 4" diameter Koch HFK328 UF membranes. Three diafiltration steps, each of VCF 2, were then carried out after which the permeate conductivity was 2 mS. A final concentration stage was then carried out to lift the retentate refractive index to 16°Br.

The GMP retentate (150 L) was spray dried on an Anhydro Integral Fluid Bed drier to give 26kg of powder containing 82.6% protein (dry basis). The PHE content of the powder was 0.24% of protein.

INDUSTRIAL APPLICABILITY

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The process of the present invention is capable of being applied to the mass production of GMP having a PHE content of not greater than about 0.5% (w/w), such purified GMP being suitable for use in the manufacture of low PHE diets for PKU patients.

CLAIMS

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- 1. A process for the purification of GMP characterised in that the purified GMP has an amino acid composition containing not greater than about 0.5%(w/w) phenylalanine (PHE) comprising:
 - contacting a GMP containing feedstock with a first anion exchanger under conditions to adsorb said GMP.
- eluting said adsorbed GMP from said anion exchanger and removing impurities from said GMP containing eluate by either:
 - i) contacting said GMP containing eluate with a cation exchanger in conditions under which said impurities in said eluate are adsorbed onto said cation exchanger, or
 - ii) precipitating said impurities in said GMP containing eluate using conditions by means of which said GMP remains in solution, or
- 20 iii) contacting said GMP containing eluate with a second anion exchanger in conditions under which said impurities in said eluate are adsorbed onto said anion exchanger and,
- recovering said GMP from whichever one or more of said steps (i), (ii) or (iii) was used.
 - 2. A process according to claim 1, characterised in that said conditions in said first anion exchanger are such that said GMP is adsorbed selectively.
- 30 3. A process according to claim 1, characterised in that said conditions in said first anion exchanger are such that both said GMP and other whey proteins in said feedstock are adsorbed and then said first anion exchanger is eluted under conditions which selectively elute said whey proteins and then elute GMP.
- A process according to any preceding claim, characterised in that said conditions for eluting said whey proteins or said impurities is obtained by adjusting said first

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anion exchanger to a pH of 4-5 if necessary and then using an appropriate eluent such as 20-60 mM sodium chloride.

- 5. A process according to any preceding claim, characterised in that the feedstock for said first anionic exchanger is a cheese or rennet whey, a UF retentate or a whey protein concentrate produced from a cheese or rennet whey, or a whey protein isolate produced from one of these by an anion exchanger.
- 6. A process according to claim 5, characterised in that said feedstock has been pretreated to remove substantially all the whey proteins contained therein except GMP.
 - 7. A process according to claim 6, characterised in that said whey proteins have been substantially removed by either heat treating said feedstock or contacting it with a cation exchanger.
 - 8. A process according to any one of claims 1-4, characterised in that said feedstock is derived from a casein or caseinate product prepared by acid precipitation and has been treated with a suitable enzyme to release GMP in solution and precipitate casein.
 - 9. A process according to any preceding claim, characterised in that said GMP containing eluate from said first anionic exchanger is treated by said alternative (i).
- 25 10. A process according to any one of claims 1-8, characterised in that said GMP containing eluate from said first anionic exchanger is treated with said alternative (ii).
- A process according to any one of claims 1-8, characterised in that said GMP containing eluate from said first anionic exchanger is treated with said alternative (iii).
- 12. A process according to any one of claims 1-8, characterised in that said GMP containing eluate from said first anionic exchanger is treated with said alternative (ii) followed by treatment with said alternative (ii).

- 13. A process according to any one of claims 1-8, characterised in that said GMP containing cluate from said first anionic exchanger is treated with said alternative (ii) followed by treatment with said alternative(i).
- A process according to any one of claims 1-8, characterised in that said GMP containing eluate from said first anionic exchanger is treated with said alternative (i) followed by treatment with said alternative (iii).
- 15. A process according to any one of claims 1-8, characterised in that said GMP containing eluate from said first anionic exchanger is treated with said alternative (iii) followed by treatment with said alternative (i).
 - 16. A process according to any preceding claim, characterised in that said GMP containing feedstock is contacted with said first anionic exchanger at a pH between 3 and 9.
 - 17. A process according claim 16, characterised in that said pH is between 3.5 and 5.
- A process according to any preceding claim, characterised in that said eluent for desorbing said GMP from said first anionic exchanger contains salt in a sufficient quantity to make the solution approximately 20-200mM and acid in a sufficient quantity to lower the pH to 2-3.
- 19. A process according to any one of claims 1-10, 12, 13, 16 and 17 characterised in that said eluent for desorbing said GMP from said first anion exchanger contains salt in a sufficient quantity to make the solution about 175mM and base in sufficient quantity to raise the pH to about 8 to 9.
- A process according to any one of claims 1-9 and 12-18, characterised in that said GMP containing eluate in alternative (i) is contacted with said cation exchanger at a pH between about 1.5 and about 4.5 and at a salt concentration of up to 100mM.
- 21. A process according to any one of claims 18-20, characterised in that said salt is sodium chloride.

- 22. A process according to any preceding claim, characterised in that said GMP is neutralised, concentrated by ultrafiltration and/or diafiltration prior to drying.
- A process according to any one of claims 1-8, 10, 12 and 13-17, characterised in that in said alternative (ii) said conditions are pH 3.5-4.5 attained by adding an acid or base to the GMP containing eluate.
 - 24. A process according to claim 23, characterised in that said acid is hydrochloric acid.
 - 25. A process according to claim 23, characterised in that said base is sodium hydroxide.
- A process according to any one of claims 23-25, characterised in that after said precipitation step the mixture is centrifuged or microfiltered and the supernatant or permeate, is either further processed according to alternative i) or is immediately neutralized, concentrated and dried to produce a GMP powder.
- A process according to any one of claims 1-8, 11 and 14-18, characterised in that in said alternative iii), said second anion exchanger is substituted with amino groups bearing at least one hydrophobic group.
- 28. A process according to claim 27, characterised in that said at least one hydrophobic group is selected from the group consisting of aliphatic and aromatic C₅-C₁₂ hydrocarbon groups.
 - 29. A process according to claim 28, characterised in that said hydrophobic group is a C₃ hydrocarbon group.
- 30. A process according to any one of claims 1-8, 11, 14-18 and 27-29, characterised in that the GMP loaded first anion exchanger in the initial step is washed at pH 4-5 with a dilute salt solution prior to GMP elution.
- 31. A process according to claim 30, characterised in that the dilute salt solution is 50 mM sodium chloride.

32. A process according to any one of claims 1-8, 11, 14-18 and 27-31, characterised in that in said alternative iii), said GMP containing eluate is adjusted to a salt concentration of 40-200 mM with a physiologically acceptable salt and the pH is at least 7.

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33. A process according to claim 32, characterised in that said salt is sodium chloride.

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34. A process according to claim 32 or 33, characterised in that the non-adsorbed GMP passing through said second anion exchanger is either further processed according to alternative i), or is immediately neutralised concentrated by ultrafiltration and/or diafiltration and dried.

35. A process according to any one of claims 1-8 and 10-34, characterised in that the GMP from the initial stage or alternative i) is concentrated by ultrafiltration before conducting alternative ii) or iii).

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36. A process according to any preceding claim, characterised in that the process is conducted under conditions which promote a high yield.

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37. A process for the purification of GMP characterised in that the purified GMP has a PHE content of not greater than about 0.5%(w/w) which comprises:

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contacting a GMP feedstock with an anion exchanger under conditions to adsorb selectively said GMP, and

eluting selectively said adsorbed GMP from said anion exchanger leaving impurities on said anion exchanger, and

recovering GMP containing eluate.

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38. A process according to claim 37, characterised in that said GMP loaded anion exchanger is washed with a dilute salt solution to remove impurities therefrom prior to said GMP elution.

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39. A process according to claim 38, characterised in that said dilute salt solution is 50 mM sodium chloride.

- A process according to any one of claims 37-39, characterised in that said anion exchanger is regenerated by eluting said impurities therefrom after said GMP elution.
- A process according to any one of claims 37 to 40, characterised in that said GMP is selectively eluted with an eluent having a pH of about 8-9 and a salt concentration of about 175mM.
- Purified GMP characterised in that the GMP is purified to contain no more than 0.5%w/w of PHE whenever prepared by a process as claimed in any preceding claim.
- 43. Purified GMP characterised in that the GMP is purified to contain no more than 0.4%w/w of PHE whenever prepared by a process as claimed in any one of claims 1-41.

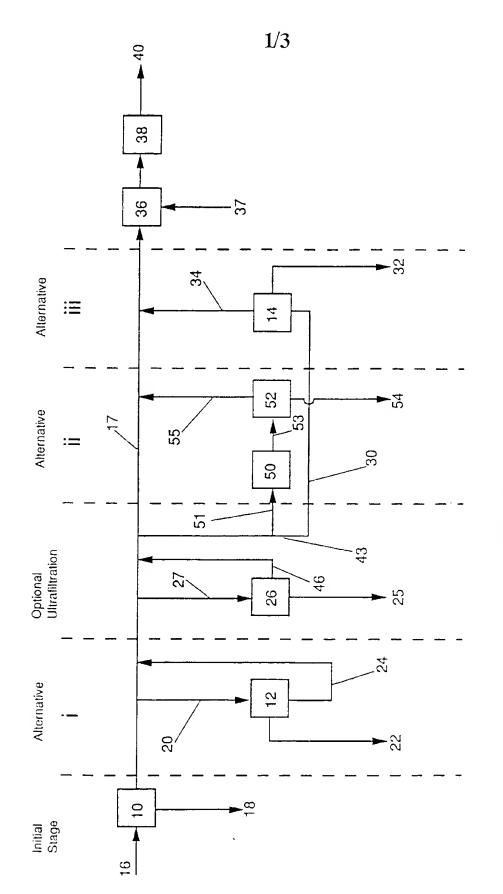


Figure 1

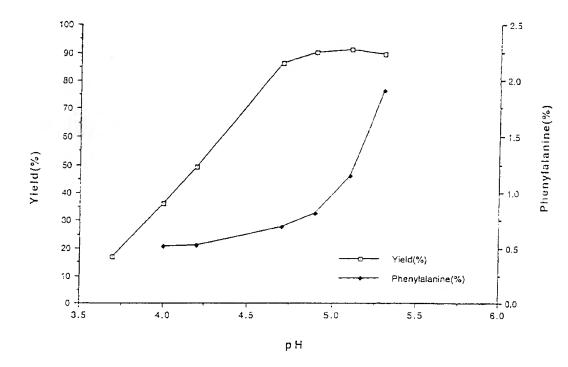


Figure 2



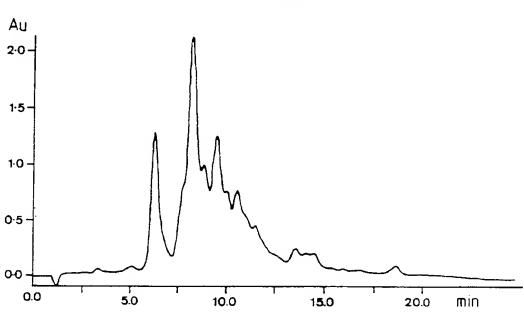


Figure 3

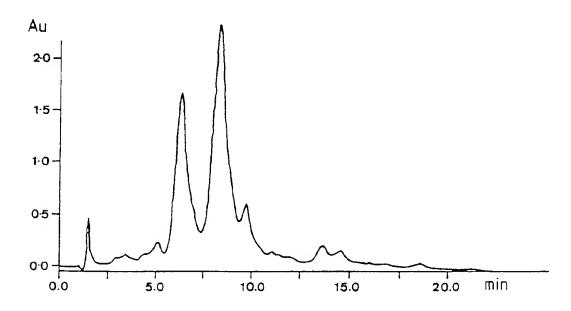


Figure 4

INTERNATIONAL SEARCH REPORT

International Application No. PCT/NZ 97/00125

A.	CLASSIFICATION OF SUBJECT MATTER		
Int Cl ⁶	A23J 1/20 3/10		
According to	International Patent Classification (IPC) or to both	national classification and IPC	
В.	FIELDS SEARCHED		
Minumum docu IPC A23J 1/	mentation searched (classification system followed by c 20 3/10	lassification symbols)	
Documentation AU: IPC as	searched other than minimum documentation to the exabove	tent that such documents are included in t	he fields searched
Electronic data Chemical Ab	base consulted during the international search (name of ostracts on STN. Keywords: Glycomacropeptic	f data base and, where practicable, search de, Casein, Macropeptide, Anion, E	terms used) exchange.
C.	DOCUMENTS CONSIDERED TO BE RELEVANT		
Category*	Citation of document, with indication, where ap	propriate, of the relevant passages	Relevant to claim No.
A	Milchwissenschaft, Volume 47, No. 11, 1992 (Met al., "Determination of Kappa-Casein glycomac without trichloroacetic acid pretreatment", pages whole document	rropeptide liquid chromatography	1-43
A	Milchwissenschaft, Volume 50, No. 10, 1995 (No. 11, 1995 (No. 11, 1995) (No. 12,	n macropeptide from cheese whey s 570 to 574 l and 3	1-43
A	"Large-scale Preparation of Kappa-Casein Glycopages 140 to 141 whole document	omacropeptide from Rennet Whey",	1-43
X	Further documents are listed in the continuation of Box C	See patent family an	nex
* Special categories of cited documents: "A" document defining the general state of the art which is not considered to be of particular relevance "E" earlier document but published on or after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention cannot be considered to involve an inventive step when the document is taken alone document of particular relevance, the claimed invention cannot be considered to involve an inventive step when the document is taken alone document of particular relevance, the claimed invention cannot be considered to involve an inventive step when the document is document of particular relevance, the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art document member of the same patent family			
	tual completion of the international search	Date of mailing of the international sear 2 1 JAN 1998	ch report
Name and mar	ling address of the ISA/AU N INDUSTRIAL PROPERTY ORGANISATION	Authorized officer GAVIN THOMPSON Telephone No.: (02) 6283 2240	17

INTERNATIONAL SEARCH REPORT

International Application No.
PCT/NZ 97/00125

C (Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT			
Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.	
	Milchwissenschaft, Volume 51, No. 6, 1996 (Munich, Germany), Chu, L. et al, "Isolation of glycomacropeptide from sodium caseinate hydrolysate solution by ultrafiltration".		
A	pages 303 to 306 see Table 2 page 305 in particular	1-43	
	European Journal of Biochemistry, Volume 50, 1975, Soulier, Solange et al. "Purification des caseines x de brebis", pages 445-452		
A	see Table 3 page 450 which shows phenylalanine in casein from 2.4 to 2.7% w/w	1-43	